

Determination of the Absolute Configuration of Ginnol, a Long-chain Aliphatic Alcohol,
by Use of a New Chiral Anisotropic Reagent

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The absolute configuration of ginnol, 10-nonacosanol, has been determined by use of a new chiral anisotropic reagent, 2NMA (2-naphthylmethoxyacetic acid), which results in the marked upfield shifts of the protons located on the same side of the aromatic ring.

Ginnol, 10-nonacosanol (**1**)¹⁾, is a component of wax produced by higher plants. Its structure has been determined by mass spectrum and synthesis.²⁾ Its stereochemistry, however, has not been elucidated yet. As in the case of **1**, the absolute configurations of many similar aliphatic alcohols have remained undetermined, because there have been no good methods to establish the configurations of such a hydroxy group that is included in a long aliphatic chain, especially when the alcoholic group is located deep in the middle of the chain. We herein describe the elucidation of the absolute configuration of **1**, mp 81.0-82.0°, [α]_D +1.7° (c=1.50, CHCl₃), obtained from the pulverized dry leaves of *Ginkgo biloba* L.

We have previously demonstrated the versatility of new chiral anisotropic reagents, 1NMA (1-naphthylmethoxyacetic acid), 2NMA (**2**: 2-naphthylmethoxyacetic acid), and 9NMA (9-anthranylmethoxyacetic acid),³⁾ the analogues of *O*-methyldelvic acid.⁴⁾ Among them, **2** is a useful reagent for long-chain compounds because the anisotropic effect of 2-naphthyl group reaches farther than others.^{3,5)}

Ginnol (**1**) was converted into (*R*) and (*S*)-2NMA derivatives (DCC/DMAP/CHCl₃). The ¹H-NMR spectra (600 MHz, CDCl₃) of the respective diastereomers are shown in Fig. 1. Spectrum (a) of (*R*)-2NMA derivative shows well-separated signals assignable to methylene protons from 9 to 2 positions, which must be the same side of 2-naphthyl group when the (*S*)-configuration of the 10-hydroxy group is assumed. These methylene signals are easily correlated with a terminal methyl (1-CH₃) by analyzing the COSY spectrum. The terminal methyls show the signals separately, one at δ 0.87 (1: upfield shift) and the other at δ 0.88 (29: normal shift). On the contrary, the chemical shifts of the two methyls are the same in the (*S*)-2NMA diastereomer since the 2-naphthyl group directs toward the longer-chain side, but 29-CH₃ is too far to be affected by the anisotropy. Yet there are observed the remarkable upfield shifts of the methylene protons from 11 to 17 positions. The $\Delta\delta$ values, which verify the (*S*)-configuration of the hydroxy group, are shown in **1a**. It is noteworthy that only one diastereomer was necessary to determine the absolute configuration: For the long-chain aliphatic alcohols, when one (within 9 bonds from 2NMA-oxycarbon) of the methyl signals shifts upfield in the (*R*)-2NMA diastereomer, the configuration of the hydroxy group is *S* (*vice versa*).

The different chemical shifts of the two methyls are still detectable in (*R*)-2NMA-1 even at 200 MHz (Fig. 1c). The conventional MTPA method⁶⁾ was not applicable to 1 because of the smaller anisotropy of the phenyl group of MTPA.

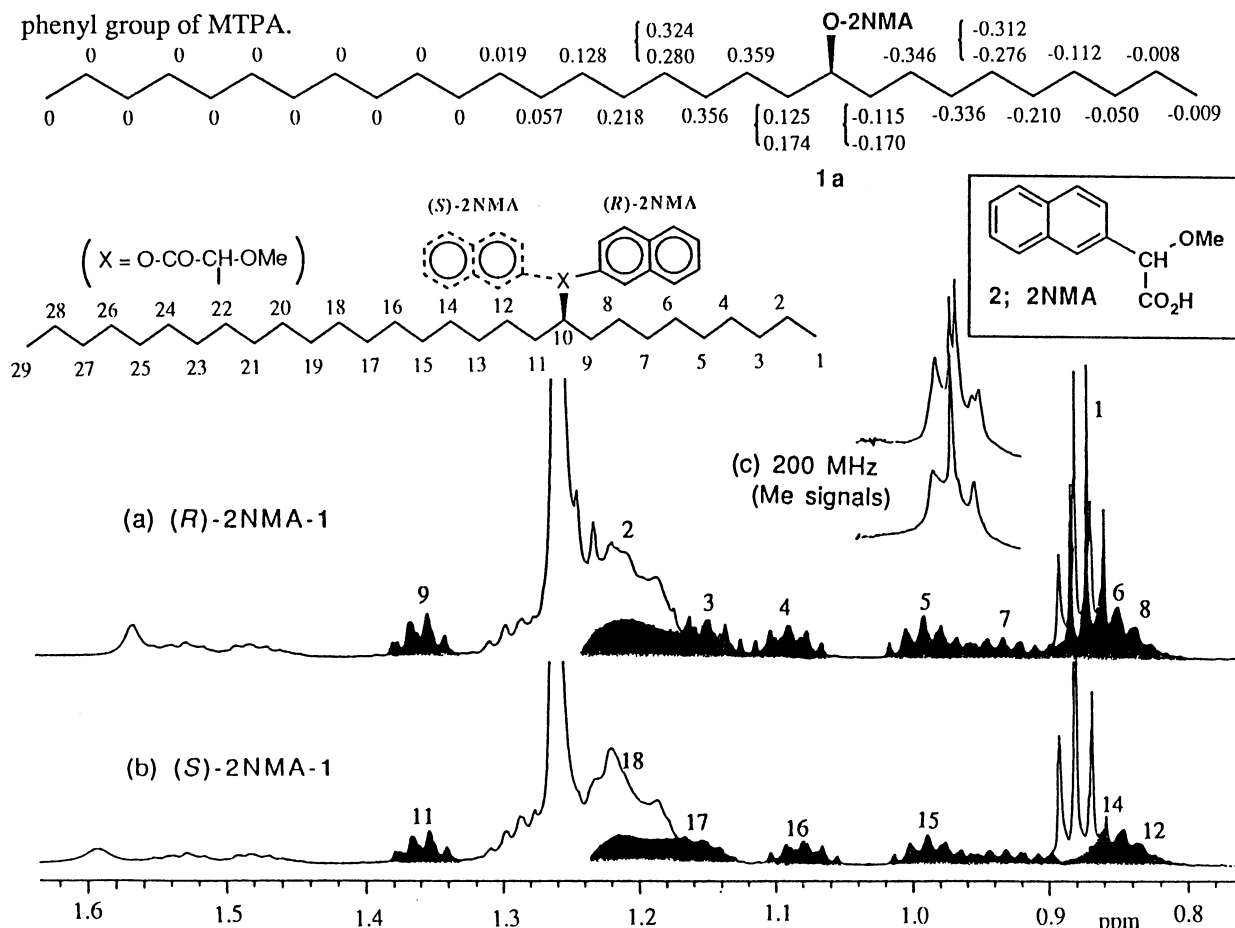


Fig. 1. Parts of the ¹H-NMR spectra (600 MHz, CDCl₃) of (a) (*R*)-2NMA-1 and (b) (*S*)-2NMA-1. (c) The methyl signals of (*R*) (above) and (*S*) (below)-2NMA-1 measured at 200 MHz.

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